# LIQUID CAUSTIC SODA – DILUTION

### DOC-1110-0005

Liquid caustic soda 50% freezes at + 12 °C. Diluting this product with softened water could prevent it from freezing: 20% or 25% solutions are the most economic as their respective freezing points are - 25 °C and - 18 °C.

# **Principle**

The liquid caustic soda 50% inlet pipe is connected to the softened water pipe with a T-piece. The calculated amount of softened water and liquid caustic soda 50% are added simultaneously. The amount of softened water is controlled by a flow meter linked to an automatic shut-off device. A static mixer placed in the pipe after the T-piece minimises temperature gradients in the liquid and produces a homogeneous solution. One-way valves should be inserted in both feed lines.

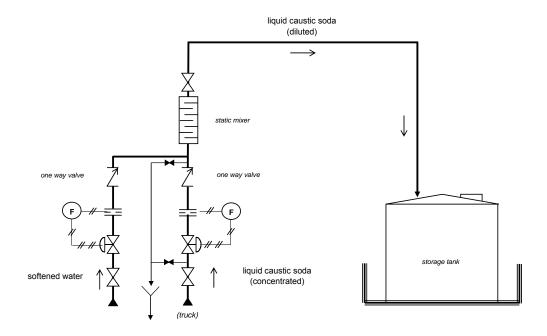
Liquid caustic soda dilution is a strong exothermic process, i.e. product temperature increases instantaneously when water is added: the product could boil and splatter, which could damage the tank.

Liquid caustic soda can be diluted safely by following these precautions:

- choose the appropriate storage tank material (i.e. check the maximum allowed temperature),
- do not dilute in storage tank,
- use softened water with a temperature of approximately 15 °C,
- insert a static mixer after the T-piece,
- adapt the flow rates to the characteristics of the process equipment (i.e. static mixer)

Please consult our safety data sheet.

# Flow Chart



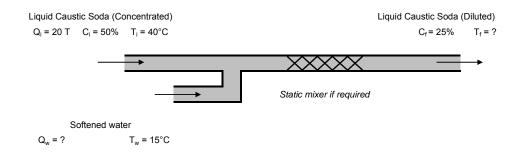
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# Calculation (example)



What is the required softened water amount - Qw?

$$Q_w = Q_i \times \frac{(C_i - C_f)}{C_f}$$

With  $Q_i = 20$  tons,  $C_i = 50\%$ ,  $C_f = 25\%$ 

$$Q_w = 20 \text{ tons } x (50\% - 25\%) = 20 \text{ tons}$$

Note:  $Q = D \times V$ , where D(\*) is density and V is liquid caustic soda volume in liters.

(\*) as per Liquid Caustic Soda - Density table

 $Q_w = 20 \text{ tons}$ 

### What is the final temperature - T<sub>f</sub>?

Reagents enthalpy:

20 ton NaOH 50% at 40 °C (\*\*) +343,9 kJ kg<sup>-1</sup> x 20000 kg = + 6878 MJ 20 ton Water at 15 °C (\*\*) +63,2 kJ kg<sup>-1</sup> x 20000 kg = + 1264 MJ Total +8142 MJ

Final solution enthalpy:

 40 ton NaOH 25% at 40 °C (\*\*)
  $-145.5 \text{ kJ kg}^{-1} \text{ x } 40000 \text{ kg} =$  -5820 MJ 

 Liberated heat:
 8142 MJ - 5820 MJ = + 2322 MJ 

 Temperature increase (\*\*\*):
  $2322 \cdot 10^3 \text{ kJ/ } (3.625 \text{ kJ kg}^{-1} ^{\circ}\text{C}^{-1} 40000 \text{ kg})$   $+ 16 ^{\circ}\text{C}$  

 Final temperature:
  $40 ^{\circ}\text{C} + 16 ^{\circ}\text{C}$   $+ 56 ^{\circ}\text{C}$ 

(\*\*) As per Liquid Caustic Soda – Enthalpy table

(\*\*\*) As per Liquid Caustic Soda – Specific Heat table

 $T_f \sim 56 \,^{\circ}C$ 

20 tons water are required to dilute 20 tons liquid caustic soda from 50% to 25% and 56 °C would be the approximate final liquid temperature.

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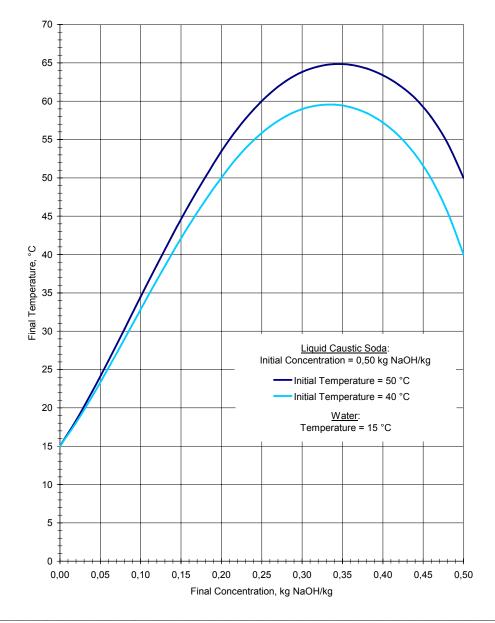
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IQUID CAUSTIC SODA

# **LIQUID CAUSTIC SODA – DILUTION**

## Liquid Caustic Soda - Dilution with Water Final Temperature



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-IQUID CAUSTIC SODA

# **LIQUID CAUSTIC SODA - ENTHALPY**

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									Tem	Temperature,	ွ ့								
0	10	20	30	40	20	09	20	80	06	100	110	120	130	140	150	160	170	180	190
0,0	41,8	84,6	126,8	168,9	211,1	254,6	299,5	341,5	383,0	424,6	466,8	508,7	9,033	592,7	635,0	677,5	720,1	763,1	806,4
2, 7	6,14	82,3	123,0	163,8	204,6	245,6	287,3	328,4	369,5	410,6	452,0	493,4	534,8	576,4	618,1	660,0	702,1	744,4	786,9
4 0	39.7	78,7	119,2	28,8	198,5	237,5	2/6/5	307.0	347.2	388,2	439,0	4/9,8	520,8	549.0	589.7	630.5	671.4	712.4	753.6
0,0	37,9	74,2	111,9	149,9	187,9	224,3	259,6	298,7	338,5	378,3	417.8	457,8	497,8	538,0	578,1	618,4	658,7	699,2	739,8
-1,3	36,0	71,6	108,8	146,1	183,6	219,2	253,4	292,0	331,4	370,8	409,7	449,3	488,9	528,6	568,3	0,809	647,8	8,789	727,8
-2,7	34,0	69,3	106,1	143,0	180,1	215,2	248,7	286,8	325,8	364,8	403,3	442,5	481,6	520,8	560,1	599,4	638,7	678,1	717,6
-3,9	32,2	67,5	104,0	140,6	177,4	212,3	245,5	283,2	321,8	360,4	398,5	437,2	476,0	514,8	553,6	592,4	631,3	670,2	709,1
-4,9	30,8	66,3	102,6	139,1	175,7	210,7	243,8	281,1	319,4	357,6	395,3	433,6	472,0	510,3	548,7	587,1	625,5	663,9	702,4
-5,3	30,1	629	102,1	138,5	175,0	210,2	243,5	280,6	318,5	356,4	393,7	431,6	469,6	507,5	545,5	583,5	621,4	659,4	697,3
-5,1	30,2	66,4	102,5	138,9	175,4	211,0	244,8	281,6	319,1	356,6	393,6	431,2	468,8	506,3	543,9	581,4	619,0	656,5	694,0
-3,9	31,3	6,79	104,1	140,5	177,0	213,1	247,5	284,1	321,3	358,5	395,2	432,4	469,6	206,7	543,9	581,0	618,2	655,3	692,3
-1,5	33,6	70,7	106,9	143,3	179,9	216,6	251,7	288,1	325,0	361,8	398,2	435,1	471,9	208,7	545,4	582,2	618,9	9'259	692,3
2,2	37,4	74,8	111,1	147,6	184,2	221,5	257,5	293,7	330,2	366,6	402,9	439,3	475,7	512,2	548,6	584,9	621,3	9,759	8,569
7,4	42,8	80,5	116,7	153,3	190,0	227,8	264,7	300,7	336,9	373,0	409,0	445,1	481,1	517,2	553,2	589,2	625,2	661,1	6,969
	20,0	8,78	124,0	160,6	197,3	235,6	273,3	309,2	345,1	380,8	416,6	452,3	488,0	523,7	559,4	595,0	9,089	1,999	9,107
	59,3	6'96	133,0	169,5	206,2	244,9	283,5	319,3	354,7	390,2	425,7	461,1	496,4	531,7	9,295	602,3	637,5	672,7	8,707
	70,8	107,9	143,9	180,3	216,9	255,8	295,2	330,7	365,9	401,0	436,3	471,3	506,3	541,2	576,1	611,0	642,9	680,7	715,4
		121,1	156,9	193,0	229,4	268,4	308,3	343,7	378,5	413,2	448,3	482,9	517,6	552,1	2,985	621,2	655,7	690,2	724,6
		136,6	171,9	207,7	243,7	282,6	322,9	358,1	392,5	426,9	461,8	496,0	530,3	564,5	298,7	632,9	0,799	701,1	735,1
		154,4	189,2	224,4	260,1	298,5	339,0	374,0	408,1	442,1	476,6	510,5	544,4	578,3	612,1	642,9	2,629	713,4	747,1
		174,8	208,9	243,5	278,5	316,1	356,5	391,3	425,0	458,7	492,9	526,5	260,0	593,5	656,9	660,3	693,7	727,1	760,4
		197,9	231,1	264,8	299,0	335,6	375,6	410,0	443,4	476,6	510,6	543,8	6,929	610,0	643,1	676,1	709,2	742,2	775,1
		223,9	256,0	288,6	321,8	356,9	396,1	430,2	463,1	496,0	529,7	562,4	595,2	652,9	9,099	693,3	725,9	758,5	791,1
		252,9	283,6	314,9	346,9	380,0	418,0	451,8	484,3	516,8	550,1	582,5	614,8	647,1	679,5	711,7	744,0	776,2	808,4
		285,1	314,1	343,9	374,5	405,1	441,5	474,8	6'909	539,0	571,9	8,509	635,8	2,799	9,669	731,5	763,4	795,2	827,0
		320,5	347,7	375,6	404,5	432,2	466,4	499,1	530,9	562,6	594,9	626,5	0,859	9,689	721,1	752,5	784,0	815,4	846,8
			384,4	410,2	437,1	461,3	492,8	524,9	556,2	587,5	619,4	650,5	681,6	712,7	743,8	774,8	802,8	836,9	6,798
				447,8	472,4	492,5	520,6	552,1	583,0	613,8	645,1	675,8	706,4	737,1	7,797	798,3	828,9	859,5	890,1
					510,4	525,7	549,9	580,6	611,0	641,4	672,0	702,3	732,5	762,7	792,9	823,1	853,2	883,3	913,4
						561,1	280,7	610,5	640,5	670,4	200,3	730,1	759,8	9,687	819,3	849,0	9,878	908,3	6,756
						598,7	613,0	641,8	671,3	9,007	729,8	759,1	788,4	817,6	846,9	876,1	905,2	934,4	963,5
						638,5	646,7	674,4	703,4	732,2	760,5	789,4	818,1	846,9	875,6	904,3	933,0	961,6	990,2
							681,8	708,4	736,8	765,1	792,5	820,8	849,1	877,3	905,5	933,7	961,8	6'686	1017,9
							718,4	743,7	771,5	799,3	825,6	853,4	881,2	6'806	936,5	964,1	991,7	1019,2	1046,6
							756,5	780,3	9,708	834,8	0'098	887,2	914,4	941,6	9,896	2,566	1022,6	1049,5	1076,4
							0,967	818,2	844,9	871,5	895,5	922,2	948,8	975,4	1001,8	1028,3	1054,6	1080,9	1107,0
							837,0	857,5	883,5	909,5	932,2	958,3	984,3	1010,3	1036,1	1061,9	1087,6	1113,2	1138,7
								898.0	923.4	948.7	970.0	995.5	10209	1046 2	1071 4	1096.5	1121 5	11161	11710
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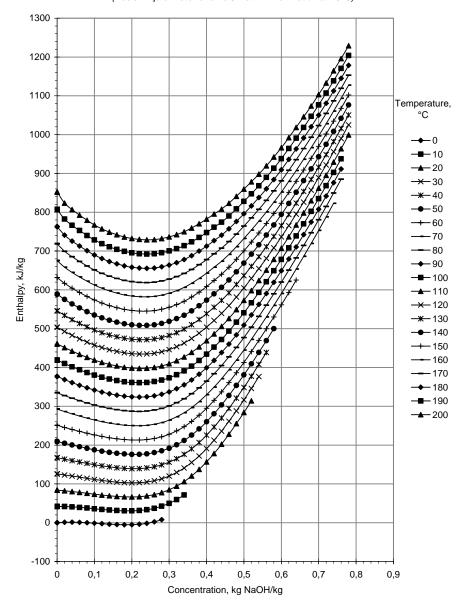
Relaterances: McCabe, Trans, A.I.Ch.E. 31, 129 (1935) - Berretti' et McCabe, Ind. Eng. Chem. 28, 247 (1936) - Berretti et McCabe, Ind. Eng. Chem. 34, 558 (1942)

LIQUID CAUSTIC SODA - ENTHALPY

# LIQUID CAUSTIC SODA - ENTHALPY

### **Enthalpy of Aqueous Solutions of NaOH**

(Basis: Liquid Water and NaOH at Infinite Dilution at 20°C)



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LIQUID CAUSTIC SODA - ENTHALPY

# LIQUID CAUSTIC SODA - SPECIFIC HEAT

# LIQUID CAUSTIC SODA - SPECIFIC HEAT

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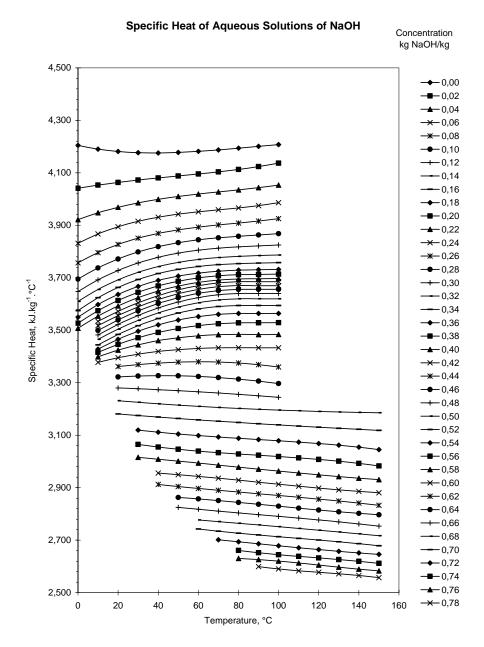
					Spec	ific He	at of A	snoənb	Solut	Specific Heat of Aqueous Solutions of NaOH, kJ.kg <sup>-1</sup> °C <sup>-1</sup>	NaOH	, kJ.kg	٠- ٢-ي							
Temperature,									Conce	Concentration, kg NaOH/kg	, kg NaC	OH/kg								
ပ္	00'0	0,02	0,04	90'0	90'0	0,10	0,12	0,14	0,16	0,18	0,20	0,22	0,24	0,26	0,28	0;30	0,32	0,34	96,0	92,0
0	4,204	4,041	3,921	3,831	3,757	3,695	3,647	3,610	3,575	3,550	3,527	3,508								
10	4,190	4,053	3,947	3,867	3,796	3,738	3,692	3,656	3,624	3,598	3,574	3,554	3,534	3,517	3,499	3,482	3,465	3,444	3,428	3,416
20	4,181	4,063	3,969	3,894	3,827	3,772	3,728	3,692	3,663	3,637	3,613	3,593	3,572	3,556	3,539	3,522	3,504	3,484	3,465	3,446
30	4,177	4,072	3,985	3,915	3,851	3,798	3,756	3,721	3,693	3,667	3,645	3,624	3,605	3,589	3,573	3,556	3,538	3,517	3,496	3,471
40	4,176	4,080	3,999	3,931	3,869	3,819	3,778	3,743	3,716	3,690	3,669	3,649	3,632	3,617	3,602	3,585	3,566	3,545	3,520	3,491
20	4,177	4,088	4,010	3,943	3,883	3,834	3,794	3,759	3,732	3,707	3,687	3,669	3,654	3,639	3,624	3,608	3,588	3,566	3,539	3,507
09	4,182	4,096	4,019	3,951	3,893	3,845	3,805	3,770	3,743	3,718	3,699	3,682	3,670	3,656	3,641	3,625	3,605	3,581	3,552	3,518
02	4,187	4,104	4,027	3,959	3,901	3,852	3,813	3,777	3,750	3,725	3,708	3,691	3,680	3,666	3,651	3,635	3,615	3,590	3,560	3,525
80	4,194	4,113	4,035	3,966	3,909	3,858	3,818	3,782	3,754	3,729	3,712	3,696	3,685	3,670	3,656	3,640	3,620	3,594	3,564	3,528
06	4,201	4,124	4,043	3,975	3,916	3,863	3,822	3,784	3,756	3,731	3,713	3,696	3,685	3,670	3,656	3,640	3,620	3,594	3,564	3,528
100	4,207	4,137	4,053	3,986	3,926	3,868	3,825	3,786	3,758	3,732	3,714	3,697	3,685	3,670	3,656	3,640	3,620	3,594	3,564	3,529
110																				
120																				
130																				
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Tomorature									Conce	Concentration, kg NaOH/kg	. kg NaC	H/ka								
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J,	0,40	0,42	0,44	0,46	0,48	0,50	0,52	0,54	0,56	0,58	0,60	0,62	0,64	0,66	0,68	0,70	0,72	0,74	0,76	0,78
0																				
10	3,398	3,378																		
20	3,424	3,395	3,362	3,322	3,280	3,231	3,181													
30	3,445	3,409	3,369	3,326	3,277	3,225	3,175	3,120	3,065	3,016										
40	3,460	3,419	3,375	3,327	3,274	3,220	3,169	3,111	3,055	3,009	2,955	2,913								
20	3,472	3,426	3,379	3,326	3,270	3,215	3,164	3,104	3,046	3,001	2,950	2,905	2,863	2,825						
09	3,479	3,431	3,380	3,324	3,266	3,210	3,158	3,098	3,039	2,994	2,943	2,897	2,857	2,818	2,777	2,743				
02	3,483	3,433	3,379	3,320	3,261	3,206	3,153	3,093	3,034	2,986	2,936	2,890	2,851	2,811	2,771	2,734	2,701			
80	3,485	3,434	3,375	3,314	3,256	3,202	3,148	3,088	3,028	2,978	2,929	2,883	2,844	2,804	2,764	2,727	2,694	2,661	2,630	
06	3,484	3,433	3,369	3,306	3,250	3,199	3,144	3,084	3,024	2,971	2,921	2,877	2,837	2,797	2,758	2,720	2,686	2,652	2,626	2,599
100	3,484	3,433	3,360	3,297	3,245	3,196	3,139	3,079	3,019	2,963	2,913	2,870	2,829	2,791	2,751	2,713	2,678	2,645	2,620	2,590
110						3,193	3,135	3,074	3,014	2,956	2,906	2,864	2,822	2,784	2,745	2,707	2,671	2,638	2,613	2,583
120						3,191	3,131	3,068	3,008	2,949	2,898	2,857	2,815	2,777	2,738	2,700	2,664	2,632	2,605	2,578
130						3,189	3,126	3,062	3,001	2,942	2,892	2,849	2,808	2,770	2,731	2,694	2,658	2,626	2,597	2,572
140						3,187	3,122	3,054	2,993	2,936	2,886	2,842	2,802	2,762	2,724	2,687	2,652	2,619	2,590	2,565
150						3,185	3,118	3,045	2,983	2,931	2,881	2,833	2,797	2,753	2,717	2,679	2,646	2,612	2,583	2,557
						Reference	s: Bertett	i & McCat	e, Ind. an	d Enginee	ing Chem	istry, 28, ;	375 (1936)	- McCabe	& Wilson	, Ind. and	Engineeri	ng Chemis	Rafarances: Bertetti & McCabe, Ind. and Engineering Chemistry, 28, 375 (1936) - McCabe & Wilson, Ind. and Engineering Chemistry, 34, 558 (1942)	8 (1942)

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# LIQUID CAUSTIC SODA - SPECIFIC HEAT



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